



Paper Type: Original Article

An Appraisal on the Dynamics of Radionuclides Contamination Matrix: A Generic Review of Radioactive Assessment in Environmental Health

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Citation:

Received: 16 June 2024

Revised: 11 September 2024

Accepted: 18 October 2024

Owunna, I. B., Ekanem, I. I., & Ikpe, A. E. (2024). An appraisal on the dynamics of radionuclides contamination matrix: A generic review of radioactive assessment in environmental health. *Trends in health informatics*, 2(1), 40-62.

Abstract

Radionuclides contamination poses a significant threat to environmental health. This contamination matrix can have detrimental effects on ecosystems, including the loss of biodiversity, disruption of food chains, environmental degradation and genetic mutations in plants and animals. In addition, exposure to radioactive contaminants can increase the risk of cancer, congenital disabilities, and other health problems in humans. This necessitated a comprehensive review of radionuclide elements, key substances, sources, properties, decay constants, decay rate, half-life, dose-response relationship, exposure pathways, measuring techniques, recorded accidents, human activities, detection and remediation techniques, policies, regulations and models on radioactive contaminants. This study employed a qualitative research approach, focusing on a review of existing literature on radionuclide contamination and assessment approaches. The literature review included peer-review articles, research reports, and publications related to radioactive assessment. Analysis of the literature aimed to identify common themes and trends in the assessment of radionuclide contamination. The findings revealed several key issues in the assessment of radionuclide contamination, including the lack of standardized assessment protocols, limited availability of data on radionuclide contamination, and the challenges in predicting the long-term effects of radioactive materials on environmental health. Moreover, it was noted that some radionuclides have half-lives of thousands of years, indicating long-term persistence that can make it difficult to remediate contaminated sites and mitigate the impacts of environmental radioactive contamination. The study suggests that addressing these issues can improve the accuracy and reliability of radioactive assessment, ultimately leading to better protection of environmental health from the impacts of radionuclide contamination.

Keywords: Radioactive assessment, Remediation techniques, Environmental health, Radionuclides contamination, Policies and regulations.

1 | Introduction

Impact assessment of radioactive contamination in the environment is a crucial aspect of environmental management, as it helps in understanding the extent of contamination, its potential effects on human health and the environment, and the effectiveness of remediation techniques [1], [2]. Radioactive contamination

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 <https://doi.org/10.22105/thi.v2i1.31>



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refers to the presence of radioactive substances in the environment, which can result from various sources such as nuclear accidents, nuclear power plants, medical facilities, and industrial activities. These substances can pose serious health risks to humans and the environment, as they can cause genetic mutations, cancer, and other adverse effects [3]. Impact assessment of radioactive contamination involves the evaluation of the extent of contamination, the pathways of exposure, the potential risks to human health and the environment, and the effectiveness of remediation techniques. The significance of impact assessment of radioactive contamination lies in its ability to provide valuable information for decision-making and risk management. By assessing the impact of contamination, policymakers, regulators, and stakeholders can make informed decisions on the need for remediation, the selection of appropriate remediation techniques, and the allocation of resources [4], [5]. This can help in minimizing the risks to human health and the environment, and in ensuring the sustainable use of natural resources. The implications of radioactive contamination are far-reaching, as they can affect ecosystems, biodiversity, food chains, and human health [6], [7]. Contaminated areas may need to be evacuated, and contaminated food and water sources may need to be restricted or treated. The long-term effects of contamination can also have social, economic, and political consequences, as affected communities may suffer from loss of livelihoods, displacement, and stigmatization [8], [9]. In terms of economic importance, impact assessment of radioactive contamination can have significant financial implications. The costs of remediation, monitoring, and long-term management of contaminated sites can be substantial and may require investments from governments, industries, and international organizations [10]. However, the economic benefits of remediation can also be considerable, as it can help in restoring the affected areas, protecting human health, and preserving ecosystems and natural resources. Impact assessment of radioactive contamination in the environment is a critical aspect of environmental management, with significant implications for human health, the environment, and the economy [11]. By conducting thorough assessments and comparative studies of remediation techniques, the risks posed by contamination can be understood better, and effective strategies for mitigating these risks can be developed. This can help in protecting human health, preserving ecosystems, and promoting sustainable development.

2| Key Milestones on Environmental Radioactive Contaminants

In recent times, the following key milestones have been achieved in the impact assessment of radioactive contamination in the environment through a comparative study of several remediation techniques:

- I. The development of advanced analytical techniques for the detection and quantification of radioactive contaminants: these techniques have significantly improved our ability to assess the extent of contamination in the environment and to track the movement of radioactive materials [12]. For example, the use of gamma spectroscopy and liquid scintillation counting has accurately aided in measuring the levels of radionuclides in soil, water, and air.
- II. The development of innovative remediation techniques for the cleanup of radioactive contamination: TRADITIONAL methods such as excavation and disposal have been replaced by more sustainable and cost-effective techniques such as phytoremediation, bioremediation, and in-situ immobilization. These techniques not only reduce the environmental impact of remediation activities but also minimize the exposure of workers to radioactive materials [13–15].
- III. The integration of risk assessment models into the Impact Assessment of Radioactive Contamination in the environment: these models allow researchers to predict the potential health risks associated with exposure to radioactive contaminants and to develop appropriate remediation strategies to mitigate these risks [16]. By incorporating factors such as dose-response relationships and exposure pathways, these models provide a comprehensive understanding of the potential impacts of radioactive contamination on human health and the environment.

The impact Assessment of Radioactive Contamination in the Environment has achieved several key milestones that have significantly shed light on the risks associated with radioactive contamination. This has brought significant progress in protecting human health and the environment from the impacts of radioactive contamination.

3 | Overview of Radionuclides

Radionuclides are unstable atoms that undergo radioactive decay, emitting radiation in the form of alpha particles, beta particles, or gamma rays. These radioactive isotopes can be found in nature, as well as being produced artificially through nuclear reactions [17]. Radionuclides play a crucial role in various fields, including medicine, industry, and research. However, their presence also raises concerns about potential health and environmental risks. One of the defining characteristics of radionuclides is their unstable nature, which leads to the emission of radiation as they decay into more stable isotopes [18], [19]. This radiation can have different properties depending on the type of decay involved. Alpha particles consist of two protons and two neutrons and are relatively large and heavy, making them less penetrating but more damaging when they interact with living tissue [20]. Beta particles, on the other hand, are high-energy electrons or positrons that are more penetrating but less damaging than alpha particles. Gamma rays are electromagnetic radiation with high energy and penetration capabilities, posing a significant risk to human health.

The presence of radionuclides in the environment can have various implications. In medicine, radioactive isotopes are used for diagnostic imaging, cancer treatment, and sterilization of medical equipment. In industry, radionuclides are used for radiography, food irradiation, and power generation in nuclear reactors [20], [21]. However, the improper handling and disposal of radioactive materials can lead to contamination of air, water, and soil, posing a risk to human health and the environment. Accidental releases of radionuclides, such as the Chernobyl and Fukushima nuclear disasters, have highlighted the potential consequences of radioactive contamination [22]. The potential health effects of radionuclides depend on factors such as the type of radiation emitted, the dose received, and the duration of exposure. Acute exposure to high doses of radiation can cause radiation sickness, organ damage, and even death. Chronic exposure to low doses of radiation may increase the risk of cancer, genetic mutations, and other long-term health effects. Therefore, it is essential to monitor and regulate the use of radionuclides to minimize the risks associated with their presence in the environment. Radionuclides are unstable atoms that emit radiation as they decay, with implications for various fields such as medicine, industry, and research [23]. While their use has brought significant benefits, the potential health and environmental risks associated with radionuclides require careful management and regulation. By understanding the properties and effects of radioactive isotopes, we can ensure their safe and responsible use in society.

4 | Radionuclide Elements and Their Properties

Radionuclide elements are atoms that have unstable nuclei and emit radiation as they decay into more stable forms. These elements play a crucial role in various fields, such as medicine, industry, and research. Understanding the different types of radionuclide elements and their properties is essential for harnessing their potential benefits while minimizing potential risks.

- I. Alpha emitters: alpha emitters are radionuclide elements that emit alpha particles during radioactive decay. Alpha particles consist of two protons and two neutrons, making them relatively heavy and low in penetrating power [24]. Examples of alpha emitters include radium-226 and plutonium-239. These elements are commonly used in smoke detectors and certain medical treatments.
- II. Beta emitters: beta emitters are radionuclide elements that emit beta particles during radioactive decay. Beta particles are high-energy electrons or positrons that have greater penetrating power than alpha particles. Examples of beta emitters include strontium-90 and iodine-131. These elements are used in medical imaging, cancer therapy, and industrial applications [25].
- III. Gamma emitters: gamma emitters are radionuclide elements that emit gamma rays during radioactive decay. Gamma rays are high-energy electromagnetic radiation with the greatest penetrating power among the three types of radiation [26]. Examples of gamma emitters include cobalt-60 and technetium-99m. These elements are widely used in medical imaging, sterilization, and industrial radiography.

- IV. Neutron emitters: neutron emitters are radionuclide elements that emit neutrons during radioactive decay. Neutrons are uncharged particles with high penetrating power, making them useful in nuclear reactors and neutron activation analysis [27]. Examples of neutron emitters include californium-252 and americium-241.
- V. Positron emitters: positron emitters are radionuclide elements that emit positrons during radioactive decay. Positrons are positively charged electrons that annihilate with electrons upon contact, producing gamma rays [28]. Examples of positron emitters include fluorine-18 and carbon-11. These elements are commonly used in Positron Emission Tomography (PET) imaging.

Radionuclide elements come in various types with unique properties that make them valuable in a wide range of applications. Understanding the characteristics of different radionuclide elements is essential for utilizing their benefits effectively while ensuring safety and minimizing risks. By following proper protocols and regulations, radionuclide elements can continue to contribute to advancements in medicine, industry, and research.

5 | Decay Constant of Radionuclide

The decay constant is a crucial parameter in the assessment of radioactive contaminants, as it plays a significant role in determining the rate at which a radionuclide decays. Radionuclides are unstable isotopes that undergo radioactive decay, emitting radiation in the form of alpha, beta, or gamma particles [29], [30]. The decay constant, denoted by the symbol λ , is a measure of the probability that a radionuclide will decay per unit time. The decay constant is defined as the reciprocal of the average lifetime of a radionuclide, which is the time it takes for half of the radioactive atoms in a sample to decay [31]. Mathematically, the decay constant is expressed as:

$$\lambda = \ln(2) / T_{1/2}, \quad (1)$$

where $T_{1/2}$ is the half-life of the radionuclide.

The half-life is the time it takes for half of the radioactive atoms in a sample to decay, and it is a characteristic property of each radionuclide. It has a direct impact on the assessment of radioactive contaminants in the environment [32]. By knowing the decay constant of a radionuclide, scientists can calculate the rate of decay and predict the concentration of the radionuclide in a given sample over time. This information is crucial for assessing the potential risks posed by radioactive contaminants to human health and the environment. The decay constant is used in the calculation of the activity of a radioactive sample, which is a measure of the rate at which radioactive decay occurs [33]. The activity of a radioactive sample is expressed in becquerels (Bq), which is defined as one decay per second. By knowing the decay constant and the initial activity of a radioactive sample, scientists can predict the activity of the sample at any given time in the future.

6 | Decay Rate of Radionuclide

Radionuclides are unstable atoms that undergo radioactive decay, emitting radiation in the form of alpha, beta, or gamma particles. The decay rate of a radionuclide is measured by its half-life, which is the time it takes for half of the atoms in a sample to decay [34]. The decay rate of radionuclides plays a significant role in assessing the impact of radioactive contaminants on human health and the environment. As radionuclides decay, they release radiation that can cause damage to living organisms, including DNA mutations, cancer, and other health problems [35]. In environmental assessments, the decay rate of radionuclides is used to calculate the concentration of radioactive contaminants in soil, water, and air. By measuring the decay rate of radionuclides in a sample, scientists can estimate the amount of radiation emitted and assess the potential impact on human health and the environment [36]. This information is crucial for developing strategies to mitigate the risks posed by radioactive contaminants and protect public health. By understanding how quickly radionuclides decay, scientists can predict how long they will remain in the environment and the potential risks they pose to human health [37]. This information is essential for making informed decisions about the management and remediation of radioactive contamination sites.

7 | Half-Life of Radionuclide

The half-life of a radionuclide is a crucial factor in the assessment of radioactive contaminants. It is defined as the time taken for half of the atoms in a sample of a radioactive substance to decay. This parameter plays a significant role in determining the potential risks associated with exposure to radioactive materials. The half-life of a radionuclide is a key characteristic that influences the behaviour of radioactive contaminants in the environment. Shorter half-lives indicate that the radionuclide decays rapidly, leading to a higher rate of radioactive emissions [38].

On the other hand, longer half-lives result in slower decay rates and prolonged exposure to radiation. The implications of the half-life of radionuclides in radioactive contaminants assessment are far-reaching [39]. A shorter half-life may result in higher levels of radiation in the immediate vicinity of a contamination source, posing a greater risk to individuals in the area. Conversely, a longer half-life may lead to the accumulation of radioactive materials in the environment over time, increasing the potential for long-term exposure and contamination of food and water sources [40]. Radionuclides with long half-lives may require specialized storage facilities to prevent the release of radiation into the environment. Equally, radionuclides with shorter half-lives may decay rapidly, reducing the need for long-term storage and monitoring [41]. When assessing the risks associated with radioactive contaminants, the half-life of radionuclides is often used as a basis for determining the level of contamination and the potential impact on human health. Regulatory agencies and environmental organizations use this information to establish safety standards and guidelines for the management of radioactive materials.

8 | Radionuclide Dose-Response Relationships

Radionuclide dose-response relationships play a crucial role in assessing the impact of radioactive contaminants on human health and the environment. These relationships provide valuable information on how different levels of exposure to radionuclides can affect living organisms and help in determining the severity of potential health risks associated with such exposure [42]. The dose-response relationship for radionuclides describes the relationship between the amount of radiation exposure and the biological response or health effects that result from that exposure. This relationship is typically expressed as a curve, with the x-axis representing the dose of radiation and the y-axis representing the likelihood or severity of health effects [43]. The impacts of radionuclide dose-response relationships can vary depending on factors such as the type of radionuclide, the route of exposure, the duration of exposure, and the sensitivity of the exposed organism. Acute exposure to high doses of radiation can cause immediate health effects such as radiation sickness, organ damage, and even death. Chronic exposure to lower doses of radiation over a long period can increase the risk of developing cancer, genetic mutations, and other long-term health problems [44]. The severity of radionuclide dose-response relationships is typically assessed using a combination of epidemiological studies, animal studies, and mathematical modelling. These studies help in estimating the risk of various health effects at different levels of radiation exposure, and determine safe exposure limits for different populations.

9 | Exposure Pathways of Radionuclides

Radionuclide exposure pathways play a crucial role in the assessment of radioactive contaminants in various environments and can vary depending on the source of contamination. Some common pathways include inhalation of radioactive particles, ingestion of contaminated food or water, and direct contact with contaminated soil or surfaces [45]. Indications of radionuclide exposure can manifest in various ways, depending on the type and level of contamination. Acute exposure to high levels of radiation can result in immediate health effects such as radiation sickness, burns, and even death [46], [47]. Chronic exposure to lower levels of radiation can lead to long-term health effects such as cancer, genetic mutations, and reproductive disorders. Radioactive contaminants can bio-accumulate in the food chain, leading to potential health risks for wildlife and humans. Contaminated water sources can pose a threat to aquatic life and crops

[48], [49]. Additionally, radioactive waste disposal sites can contaminate soil and groundwater, posing long-term risks to surrounding communities.

10 | Methods for Measuring Radionuclide Exposure

Radionuclide exposure is a significant concern in the assessment of radioactive contaminants, as it poses potential health risks to individuals and the environment. In order to accurately measure the levels of radionuclide exposure, the following methods can be employed: Some of the common methods for measuring radionuclide exposure are highlighted as follows:

- I. Through the use of dosimeters: dosimeters are devices that individuals can wear to measure the amount of radiation they are exposed to over a certain period. These devices can provide valuable data on the levels of radionuclide exposure experienced by individuals in different environments, such as workplaces or areas near radioactive waste sites. However, dosimeters may not always provide a complete picture of radionuclide exposure, as they only measure external radiation and do not account for internal exposure through ingestion or inhalation of radioactive particles [50], [51].



Fig. 1. Different types of Dosimeters [52].

- II. Through environmental monitoring: this involves collecting samples of air, water, soil, and food from areas potentially contaminated with radioactive materials and analyzing them for the presence of specific radionuclides. Environmental monitoring can provide valuable information on the levels of radionuclide contamination in different environmental media, helping to assess the potential risks to human health and the environment. However, environmental monitoring may be limited by factors such as the availability of sampling locations and the cost of analysis [53], [54].

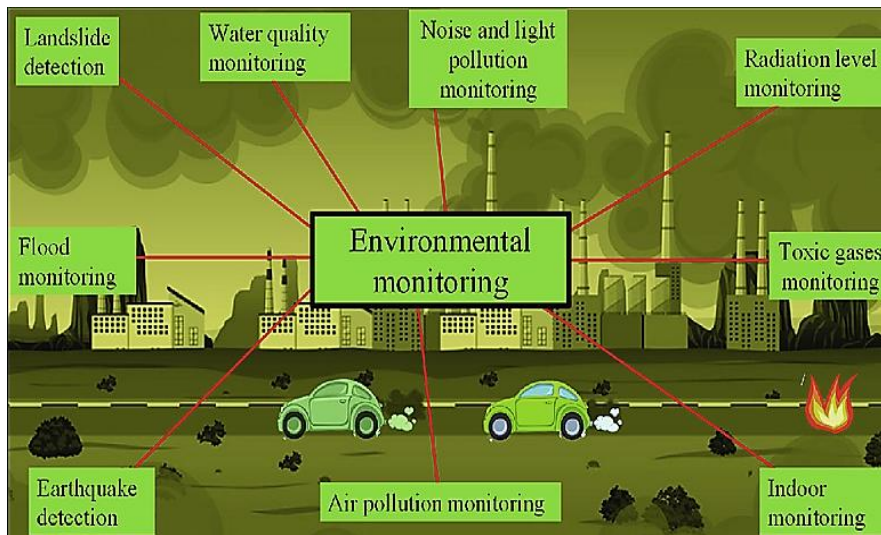


Fig. 2. Environmental monitoring [55].

- III. Through biological monitoring: this involves measuring the levels of radionuclides in biological samples, such as blood, urine, or tissue samples, taken from individuals who may have been exposed to radioactive contaminants. Biological monitoring can provide valuable information on the internal dose of radionuclides received by individuals, helping to assess the potential health risks associated with radionuclide exposure. However, biological monitoring may be limited by factors such as the availability of appropriate biomarkers for specific radionuclides and the variability in individual responses to radiation exposure [54], [56].

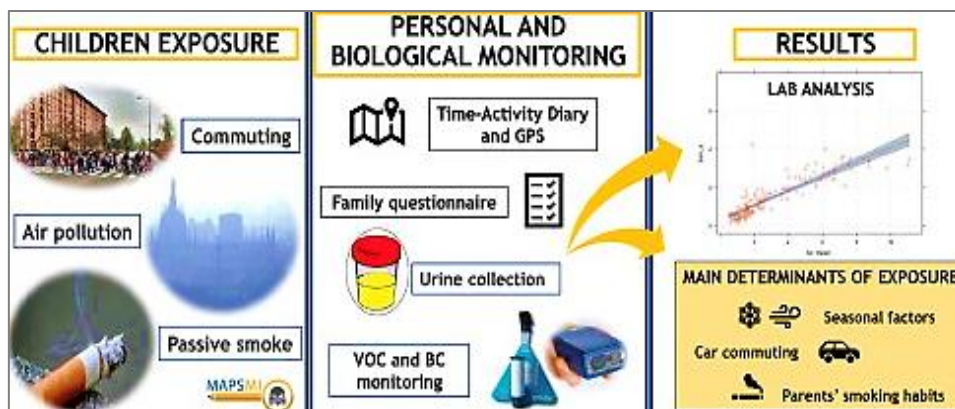


Fig. 3. Biological monitoring [57].

Measuring the levels of radionuclide exposure in radioactive contaminants assessment is crucial for understanding the potential risks to human health and the environment. Dosimeters, environmental monitoring, and biological monitoring are key methods that can be used to assess radionuclide exposure, each with its own strengths and limitations.

11 | Sources of Radionuclides

In order to assess and mitigate the risks associated with radionuclides, it is crucial to identify the sources from which they originate. Some of the sources of radionuclides in radioactive contaminants assessment are as follows:

- I. Nuclear power plants: these facilities use uranium and other radioactive materials to generate electricity, resulting in the production of radioactive waste. Accidents or leaks at nuclear power plants can release radionuclides into the environment, contaminating air, water, and soil [58].

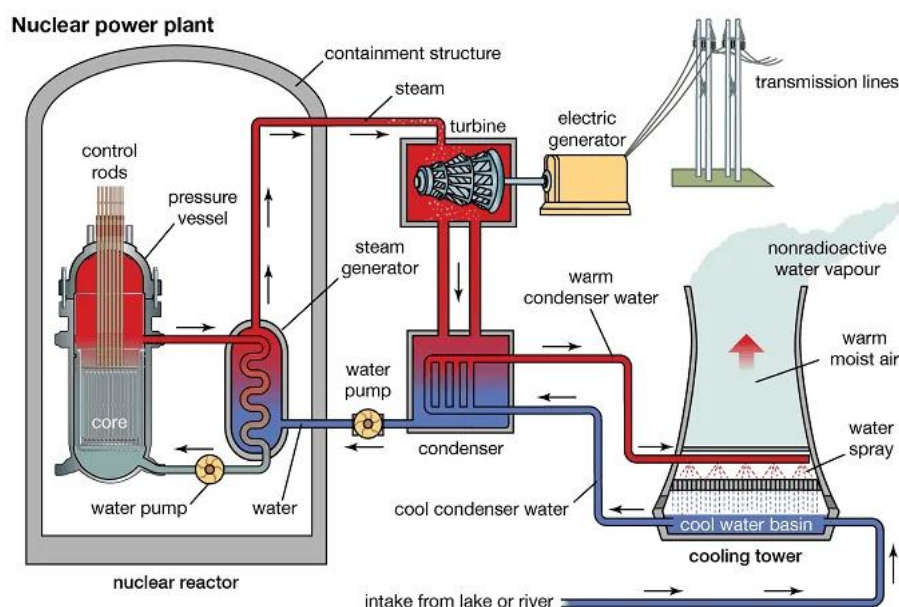


Fig. 4. Nuclear power plant [59].

- II. Medical facilities: DIAGNostic imaging procedures such as X-rays and CT scans use radioactive isotopes to visualize internal organs and tissues. Improper disposal of radioactive waste from medical facilities can lead to the contamination of water sources and soil [60].



Fig. 5. Medical facilities [61].

- III. Industrial activities: such as mining and processing of radioactive materials, also contribute to the release of radionuclides into the environment. Uranium mining, for example, can result in the dispersion of radionuclides into the air and water, posing a risk to nearby communities [62].
- IV. Natural sources of radionuclides include radon gas, which is produced by the decay of uranium in soil and rocks. Radon can seep into buildings and accumulate to dangerous levels, increasing the risk of lung cancer in occupants [63].

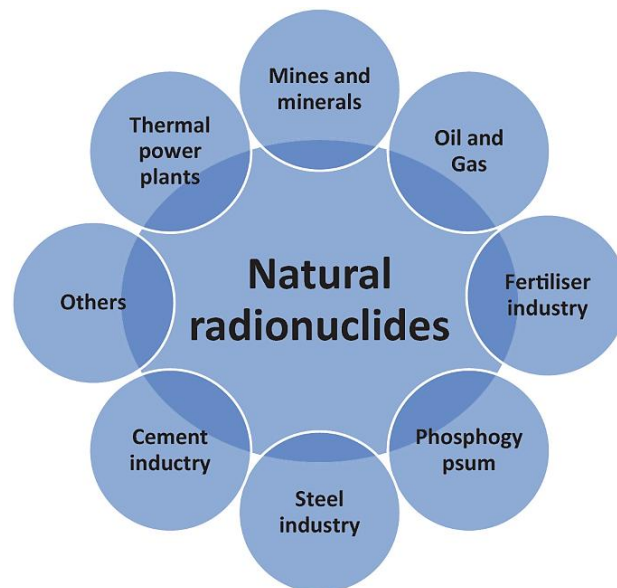


Fig. 6. Natural sources of radionuclides [64].

- V. Nuclear weapons testing: this has been a significant source of radionuclide contamination in the environment. The detonation of nuclear weapons releases a variety of radioactive isotopes into the atmosphere, soil, and water, which can have long-lasting effects on human health and the environment [65].



Fig. 7. Nuclear weapons testing [66].

- VI. Accidents involving nuclear weapons, such as the Chornobyl and Fukushima disasters, have also resulted in the release of large amounts of radioactive contaminants into the environment [66].
- VII. Disposal of radioactive waste from research facilities: this is another important source of radionuclide contamination. Radioactive waste from research facilities can include spent nuclear fuel, contaminated equipment, and other materials that have been exposed to radiation. Improper disposal of this waste can lead to the release of radioactive contaminants into the environment, posing a risk to human health and the ecosystem [67], [68].

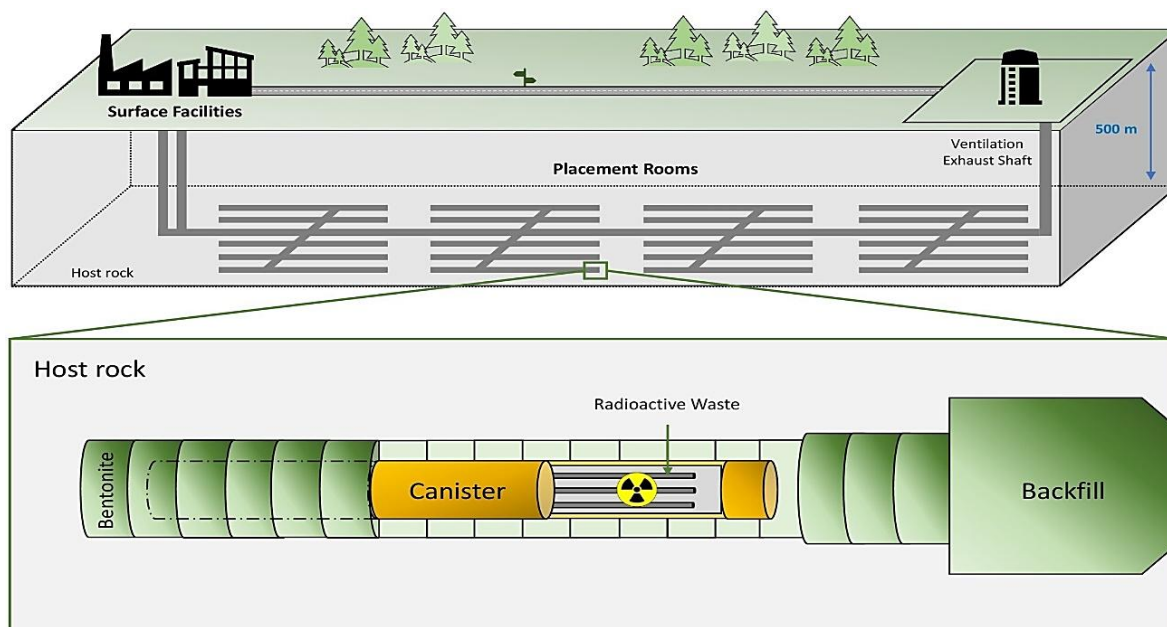


Fig. 8. Disposal of radioactive waste from research facilities [69].

The assessment of radioactive contaminants requires a thorough understanding of the sources of radionuclides. By identifying and monitoring these sources, effective measures can be implemented to protect human health and the environment from the harmful effects of radiation exposure. It is imperative that regulatory agencies, industries, and the public work together to prevent and mitigate the risks associated with radionuclides.

12 | Key Hazardous Substances Present in Radionuclide

Radionuclides contain hazardous elements that can be found in various chemical and biochemical radioactive substances, each with its unique properties and effects. These elements can also be present and contribute to the overall toxicity of radioactive materials. Some of the most well-known hazardous elements found in radionuclides are highlighted as follows:

- I. Uranium: uranium is a naturally occurring radioactive element that is commonly used in nuclear power plants and weapons. Exposure to uranium can lead to various health problems, including kidney damage, lung cancer, and congenital disabilities [70], [71].
- II. Plutonium: PLUTonium is a highly toxic and radioactive element that is used in nuclear weapons and reactors. Exposure to plutonium can cause cancer, organ damage, and other serious health issues [72], [73].
- III. Cesium-137: is a radioactive isotope that is produced during nuclear fission and is commonly found in nuclear waste. Exposure to cesium-137 can lead to radiation sickness, cancer, and other health problems [74], [75].
- IV. Strontium-90: this is another hazardous biochemical element found in radionuclides. Strontium-90 is a byproduct of nuclear fission and can accumulate in the bones, leading to bone cancer and other health issues [76].
- V. Iodine-131: this is commonly released during nuclear accidents or nuclear weapons testing. Iodine-131 is a beta and gamma emitter that can accumulate in the thyroid gland, leading to an increased risk of thyroid cancer [77].
- VI. Americium-241: is a byproduct of nuclear reactions that emits alpha particles and poses a significant risk to human health if ingested or inhaled [78].
- VII. Technetium-99: is a common radionuclide found in nuclear waste and medical imaging procedures. Technetium-99 has a long half-life and can contaminate soil and water, posing a risk to ecosystems and human health [79].
- VIII. Tritium is a radioactive form of hydrogen, which is often released from nuclear power plants and can contaminate groundwater, leading to long-term health effects if consumed [80].
- IX. Radon-222: this is a naturally occurring radioactive gas that can seep into homes and buildings from the ground, posing a risk of lung cancer if inhaled over long periods [81], [82].
- X. Thorium-232: is a radioactive element found in certain minerals and can pose a risk to workers in industries that handle thorium-containing materials [83].

While Uranium, Plutonium, Cesium-137, and Strontium-90 are well-known hazardous elements found in radionuclides, several other elements can also be present and contribute to the overall toxicity of radioactive materials. Regulators, policymakers, and industries must consider the full range of hazardous elements found in radionuclides to manage and mitigate the risks associated with radioactive materials effectively. Overall, radionuclides contain a variety of hazardous chemical and biochemical elements that can pose serious health risks to humans and the environment [84]. It is important to be aware of these hazards and take appropriate precautions to minimize exposure to these harmful substances. By understanding the risks associated with radionuclides, we can better protect ourselves and future generations from the potential dangers of radioactive contamination.

13 | Cases/Accidents Recorded for Radioactive Contaminants

While the incidents at Chernobyl and Fukushima are perhaps the most well-known cases of radioactive contamination, there have been several other incidents recorded throughout history. Cases of radioactive contamination have been recorded in various events throughout history, raising concerns about the potential health and environmental impacts of exposure to these harmful substances. They include the following:

- I. One such event occurred in 1986 when the Chernobyl nuclear power plant in Ukraine experienced a catastrophic meltdown, releasing large amounts of radioactive material into the atmosphere. This incident resulted in widespread contamination of the surrounding area, leading to long-term health effects for residents and wildlife [85].
- II. Another notable case of radioactive contamination occurred in 2011 when the Fukushima Daiichi nuclear power plant in Japan was severely damaged by a tsunami, causing multiple reactor meltdowns and the release of radioactive material into the environment. This event had far-reaching consequences, including

the evacuation of thousands of residents from the surrounding area and concerns about the long-term effects of exposure to radiation [86].

- III. Another is the Mayak Production Association in Russia, which has been responsible for numerous radioactive contamination incidents over the years. The Mayak facility, which was built in the late 1940s, has been linked to several incidents of radioactive contamination, including the Kyshtym disaster in 1957. This incident, which was caused by the explosion of a storage tank containing radioactive waste, resulted in the release of a significant amount of radioactive material into the environment, contaminating an area of over 20,000 square kilometres [87].
- IV. The Three Mile Island accident in the United States is another notable case of radioactive contamination. In 1979, a partial meltdown of a reactor at the Three Mile Island nuclear power plant in Pennsylvania resulted in the release of radioactive gases and iodine into the environment. While the accident did not result in any immediate deaths or injuries, it raised concerns about the safety of nuclear power plants and the potential for radioactive contamination [88].
- V. The Sellafield nuclear reprocessing plant, located in Cumbria, England, has been the site of several radioactive contamination accidents over the years. One of the most notable incidents occurred in 1957, when a fire broke out in the facility's plutonium processing plant, releasing a significant amount of radioactive material into the atmosphere. This incident led to widespread contamination of the surrounding area and raised concerns about the safety of the plant's operations [89].
- VI. The Hanford Site in Washington State has also experienced multiple incidents of radioactive contamination. One of the most well-known incidents occurred in 2017, when a tunnel containing radioactive waste collapsed, leading to the release of radioactive material into the environment. This incident raised concerns about the ageing infrastructure at the site and the potential risks posed by the storage of radioactive waste [90].

These incidents serve as a stark reminder of the potential dangers associated with nuclear power and the importance of strict safety regulations and oversight. These incidents highlight the need for continued vigilance and regulation in the nuclear industry to prevent future accidents and protect the environment and public health.

14 | Human Activities Responsible for Radioactive Contaminants

Human activities have been responsible for the release of radioactive contaminants into the environment, posing a significant threat to human health and the ecosystem.

- I. Activities in nuclear power plants, nuclear weapons testing, and nuclear waste disposal release radioactive materials such as uranium, plutonium, and caesium into the environment, contaminating soil, water, and air. The accidents at the Chernobyl and Fukushima nuclear power plants are stark reminders of the devastating consequences of nuclear accidents on human health and the environment [91].
- II. Activities in mining and processing release radioactive particles such as uranium and thorium into the air and water, contaminating the surrounding environment and posing a risk to human health. The improper disposal of radioactive waste from mining and processing activities further exacerbates the problem, leading to long-term contamination of the environment [92].
- III. Medical procedures involving the use of radioactive materials, such as radiation therapy and diagnostic imaging, also contribute to the release of radioactive contaminants into the environment. While these procedures are essential for diagnosing and treating various medical conditions, the improper disposal of radioactive waste from medical facilities can lead to contamination of the environment and pose a risk to public health [93].
- IV. Industrial activities such as the production of nuclear weapons, the use of radioactive materials in manufacturing processes, and the disposal of radioactive waste from industrial facilities also contribute to the presence of radioactive contaminants in the environment. These activities release radioactive materials

into the air, water, and soil, contaminating the surrounding environment and posing a risk to human health [94].

Human activities are responsible for the release of radioactive contaminants into the environment, posing a significant threat to human health and the ecosystem. Policymakers, industry stakeholders, and the public need to take proactive measures to minimize the release of radioactive contaminants and mitigate their impact on the environment. By implementing strict regulations, promoting sustainable practices, and investing in clean energy alternatives, the reliance on radioactive materials can be minimized, and the environment can be protected for future generations.

15 | Occurrence of Radionuclide Process

The occurrence of radionuclides can be categorized as natural, unavoidable, resulting from nuclear fission and thermonuclear explosions and synthetic as follows:

- I. Natural radionuclides are elements that exhibit radioactive decay in nature. These radionuclides can be classified into three main categories: primordial, secondary, and cosmogenic radionuclides. Primordial radionuclides are those that have existed since the formation of the Earth, such as uranium-238 and thorium-232. Secondary radionuclides are formed through the decay of primordial radionuclides, while cosmogenic radionuclides are produced by interactions between cosmic rays and elements in the Earth's atmosphere [53], [95].
- II. Nuclear fission is a process in which the nucleus of an atom splits into two or more smaller nuclei, releasing a large amount of radioactive energy. This process can occur naturally in radioactive materials or can be induced in a controlled manner in nuclear reactors [96]. However, thermonuclear explosions, such as those seen in nuclear weapons, involve the rapid release of energy from the fission of heavy elements like uranium or plutonium. On the other hand, the irradiation of nuclear fuel in reactors involves the controlled fission of uranium or plutonium to generate heat for electricity production [97], [98].
- III. Synthetic radionuclides are artificially produced elements that exhibit radioactive decay. These radionuclides are typically created in particle accelerators, such as cyclotrons, by bombarding stable elements with high-energy particles. Synthetic radionuclides have a wide range of applications, including medical imaging, industrial radiography, and scientific research [99].

Natural radionuclides play a significant role in the Earth's natural processes, while nuclear fission and synthetic radionuclides have important applications in energy production, medicine, and research. Understanding the occurrence and properties of these radionuclides is crucial for ensuring their safe and responsible use in various fields.

16 | Effects, Detection and Remediation Techniques of Radionuclides in the Atmosphere

When radionuclides are released into the atmosphere, either through natural processes or human activities, they can have far-reaching effects on air quality and public health. The presence of radionuclides in the atmosphere can lead to increased levels of radiation exposure for individuals living in the vicinity of the release [100]. This exposure can have serious health consequences, including an increased risk of cancer and other radiation-related illnesses. Detecting radionuclides in the atmosphere is a challenging task due to their low concentrations and the presence of background radiation. However, advances in technology have made it possible to detect even trace amounts of radionuclides in the air [101]. One common method of detection is through the use of air sampling devices, which collect air samples and analyze them for the presence of radionuclides. Other detection techniques include gamma spectroscopy and liquid scintillation counting, which can provide more detailed information about the types and concentrations of radionuclides present.

Once radionuclides have been detected in the atmosphere, remediation techniques must be implemented to reduce their impact on human health and the environment [101]. One common remediation technique is the

use of air filtration systems, which can remove radionuclides from the air before individuals inhale them. In addition, decontamination efforts may be necessary to remove radionuclides from surfaces and prevent further spread of contamination. The effects of radionuclides on the atmosphere can be significant and long-lasting [102]. By implementing effective detection and remediation techniques, the impact of radionuclides on air quality and public health can be minimized.

17 | Effects, Detection and Remediation Techniques of Radionuclide on Land/Soil

Radionuclide elements can be naturally occurring or manufactured, and their presence in the soil can lead to contamination and potential health risks for humans and wildlife. Radioactive elements such as uranium, thorium, and radium can leach into the soil and groundwater, leading to the contamination of crops and drinking water [103]. This can pose serious health risks for humans and animals that consume these contaminated resources. Additionally, radionuclides can disrupt the natural ecosystem by affecting the growth and reproduction of plants and animals. There are several methods for detecting radionuclides in soil, including gamma spectroscopy, alpha spectroscopy, and liquid scintillation counting [104]. These techniques allow for the identification of specific radioactive isotopes present in the soil and measure their concentration levels. One common method for remediation is soil washing, which involves removing the contaminated soil and treating it with chemicals to extract the radionuclides. Another technique is phytoremediation, which uses plants to absorb and accumulate radionuclides from the soil [105].

Additionally, barriers can be installed to prevent the spread of radioactive contaminants in the soil and groundwater. Detecting the presence of these radioactive elements in soil is essential for assessing the extent of contamination, while remediation techniques can be implemented to reduce their impact on the environment [106]. This can protect the environment and ensure the health and safety of future generations.

18 | Effects, Detection and Remediation Techniques of Radionuclides on Water Sources

Radionuclides are radioactive isotopes that can be found in water sources, posing a significant threat to human health and the environment. The presence of radionuclides in water can result from natural processes, such as the decay of uranium and thorium in rocks and soil, as well as human activities, including nuclear power plants, mining, and industrial processes [107]. The effects of radionuclides on water quality can be detrimental, leading to several health challenges. Various methods can be used to detect radionuclides, including gamma spectroscopy, liquid scintillation counting, and alpha spectroscopy. These techniques allow for the identification and quantification of specific radionuclides present in water, enabling authorities to take appropriate remediation measures. Remediation techniques for radionuclides in water can vary depending on the source and type of contamination. In surface water, physical methods such as filtration and sedimentation can be used to remove radionuclides [108]. In subsurface water, techniques such as pump-and-treat systems and in-situ immobilization can be employed to remediate contaminated groundwater. For drinking water sources, advanced treatment methods such as ion exchange, reverse osmosis, and activated carbon filtration can effectively remove radionuclides. Remediation of radionuclides in groundwater, which can affect large areas and persist for long periods, often involves a combination of physical, chemical, and biological methods [109]. In-situ techniques such as bioremediation and phytoremediation can be used to treat contaminated groundwater without the need for extraction and treatment. With advanced detection techniques and the implementation of appropriate remediation measures, the impact of radionuclides on water sources can be minimized, ensuring the safety and health of communities.

19 | Effects, Detection and Remediation Techniques of Radionuclides on Human Health

Exposure to radionuclides can have various detrimental effects on the human body, including genetic mutations, cancer, reproductive problems and other serious health issues. The effects of radionuclide exposure on human health are well-documented. Ionizing radiation can damage DNA, leading to mutations that may result in cancer or other health problems [35]. Additionally, exposure to high levels of radionuclides can cause acute radiation sickness, which can be fatal if not treated promptly. Chronic exposure to low levels of radionuclides can also have long-term health effects, such as an increased risk of cancer. Various techniques are used to detect radionuclides in the environment, including gamma spectroscopy, liquid scintillation counting, and alpha spectrometry [110]. These methods allow scientists to identify the presence of radionuclides in soil, water, and air, enabling them to take appropriate remediation measures. One common remediation technique is soil washing, which involves removing contaminated soil and treating it to remove the radionuclides. Another method is phytoremediation, in which plants are used to extract radionuclides from the soil [111]. Additionally, barriers can be constructed to prevent the spread of radionuclides in groundwater. Detecting and remediating radionuclide contamination is essential to protect public health and prevent the harmful effects of radiation exposure.

20 | Policies Introduced to Control Radioactive Contaminants

In order to mitigate this threat, various policies have been introduced to control and regulate the release of radioactive materials into the environment. Policies to control radioactive contaminants are as follows:

- I. The atomic energy act of 1946: this act established the Atomic Energy Commission (AEC) to regulate the development and use of atomic energy in the United States. The AEC was responsible for overseeing the production, use, and disposal of radioactive materials to ensure public safety [112].
- II. The clean air act: the clean air act regulates air emissions from sources such as nuclear power plants and other facilities that release radioactive contaminants into the atmosphere. The act sets limits on the amount of radioactive materials that can be released and requires facilities to monitor and report their emissions [113].
- III. The clean water act: the clean water act regulates the discharge of pollutants, including radioactive contaminants, into water bodies. Facilities that release radioactive materials into water must obtain permits and comply with strict regulations to protect aquatic ecosystems and human health [114].
- IV. The Nuclear Regulatory Commission (NRC) regulations: the NRC is responsible for regulating the nuclear industry in the United States. The commission sets standards for the safe operation of nuclear facilities, including the handling and disposal of radioactive materials [115].
- V. The Comprehensive Environmental Response, Compensation, and Liability Act (CERCLA): CERCLA, also known as the Superfund program, addresses the cleanup of hazardous waste sites, including those contaminated with radioactive materials. The act provides funding for the remediation of contaminated sites and holds responsible parties accountable for the cleanup [116].
- VI. The radiation control for health and safety act: this act regulates the use of radiation-emitting devices in medical and industrial settings to protect workers and the public from exposure to harmful levels of radiation. The act sets standards for the safe use of radiation and requires facilities to be licensed and inspected regularly [117].

Several policies have been introduced to control radioactive contaminants and protect public health and the environment. While these policies have made progress in regulating the release of radioactive materials, continued efforts are needed to ensure their effectiveness and address emerging challenges in the management of radioactive contaminants. Regulatory agencies, industry stakeholders, and the public must work together to uphold these policies and safeguard against the risks posed by radioactive contaminants.

21 | Regulatory Bodies on Radioactive Contaminants

Regulatory bodies play a crucial role in ensuring the safety and protection of the public from radioactive contaminants. The following bodies are responsible for setting standards, monitoring compliance, and enforcing regulations to prevent exposure to harmful radiation:

- I. Nuclear Regulatory Commission: the NRC is responsible for regulating the use of nuclear materials and facilities to ensure the safety of workers and the public. The NRC sets standards for radiation protection, conducts inspections, and enforces regulations to prevent the release of radioactive contaminants into the environment [118].
- II. Environmental Protection Agency (EPA): the EPA is responsible for setting standards for environmental quality, including limits on the release of radioactive contaminants into air, water, and soil. The EPA works closely with other federal and state agencies to monitor radiation levels and enforce regulations to protect public health [119].
- III. The International Atomic Energy Agency (IAEA) plays a key role in regulating radioactive contaminants. The IAEA sets international standards for nuclear safety and security, conducts inspections of nuclear facilities, and provides technical assistance to countries to help them comply with regulations. The IAEA also works to prevent the spread of nuclear weapons and promote peaceful uses of nuclear technology [120], [121].
- IV. Numerous state and local agencies oversee radioactive contaminants. For example, the California Department of Public Health regulates radiation sources in the state and conducts inspections of facilities to ensure compliance with regulations. Local health departments may also have jurisdiction over radioactive materials and facilities within their jurisdictions [121].

Regulatory bodies play a critical role in protecting public health from the dangers of radioactive contaminants. By setting standards, monitoring compliance, and enforcing regulations, these agencies help to ensure that nuclear materials are used safely and responsibly. These regulatory bodies must continue to work together to address the challenges of radioactive contamination and safeguard public health for future generations.

22 | Radionuclides Models for Risk Assessment

Risk assessment models for radionuclides for radioactive contaminants assessment play a crucial role in evaluating the potential health risks associated with exposure to these contaminants. The following models presented in *Table 1* have been developed to assess the risks posed by radionuclides:

Table 1. Radionuclides models for assessing the risk from radioactive contaminants.

S/N.	Radionuclides Models for Risks Assessment	Model Description
1	Dose-response model	One of the most widely used risk assessment models for radionuclides is the dose-response model, which estimates the risk of adverse health effects based on the dose of radiation received by an individual. This model takes into account factors such as the type of radionuclide, the route of exposure, and the duration of exposure to calculate the likelihood of developing health effects such as cancer or genetic mutations.
2	Probabilistic risk assessment model	Another commonly used risk assessment model is the probabilistic risk assessment model, which takes into account uncertainties in the data and assumptions used to estimate risks to provide a more realistic estimate of risk level. This model uses statistical methods to account for variability and uncertainty in factors such as the dose-response relationship and exposure pathways, providing a more comprehensive assessment of the potential risks posed by radionuclides. It estimates the likelihood of adverse health effects occurring as a result of exposure to radionuclides.

Table 1. Continued.

S/N.	Radionuclides Models for Risks Assessment	Model Description
3	Environmental risk assessment model	The environmental risk assessment model is another important tool for evaluating the risks associated with radionuclides in the environment. This model considers factors such as the bioaccumulation of radionuclides in the food chain, the potential for long-term exposure through contaminated water sources, exposure pathways to estimate the risk to ecosystems, and the impact of radionuclides on ecosystems. By taking into account these complex interactions, the environmental risk assessment model provides a more holistic understanding of the risks posed by radionuclides in the environment.
4	Biokinetic model	The biokinetic model focuses on the uptake, distribution, and elimination of radionuclides within the human body. These models help to predict the internal dose received by individuals and assess the potential health effects. By incorporating physiological parameters such as metabolism and excretion rates, the biokinetic model can provide valuable insights into the internal dose received by individuals exposed to radioactive contaminants.
5	Ecological risk assessment model	The ecological risk assessment model evaluates the potential impacts of radionuclide radioactive contaminants on ecosystems and wildlife. This model considers factors such as population dynamics, habitat quality, species interactions, bioaccumulation and biomagnification of contaminants in the food chain, as well as the potential for long-term ecological effects to estimate the risk to ecological systems.
6	Multi-pathway exposure assessment model	Multi-pathway exposure assessment model is used to assess the risks associated with combined exposure to radionuclide radioactive contaminants through multiple pathways, such as ingestion, inhalation, and dermal contact. These models help to identify the most significant exposure routes and prioritize risk management strategies.

23 | Conclusion

This study delved into the appraisal of radionuclide contamination, providing valuable insights into the dynamics of radioactive assessment approaches. Through a generic review of various methodologies and techniques used in assessing radionuclide contamination, this study identified key factors that influence the spread and impact of radioactive substances in the environment. One of the key findings from this study is the importance of considering the long-term effects of radionuclide contamination on human health and the environment. By analyzing the behaviour of radionuclides in different environmental matrices, scientists can develop models to predict the movement and accumulation of radioactive substances over time. This information is crucial for developing effective strategies for mitigating the risks associated with radionuclide contamination.

Furthermore, this study has also highlighted the need for continuous monitoring and assessment of radionuclide contamination in order to ensure the safety of populations living in affected areas. By using a combination of field measurements, laboratory analyses, and modelling techniques, scientists can accurately assess the levels of radioactive substances in the environment and determine the potential risks to human health. However, the findings from this study have provided a solid foundation for further research in this field. By building on the knowledge and insights gained, the behaviour of radionuclides in the environment can be improved upon, and more effective strategies for managing and mitigating the associated risks can be developed for future generations.

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